## A NOVEL ALDEHYDE SYNTHESIS BASED ON ANHYDRO-BASES OF THE s-TRIAZOLE SERIES Gábor Doleschall

Research Group for Alkaloid Chemistry of the Hungarian Academy of Sciences
1111 Budapest, Gellért tér 4, Hungary

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Stable anhydro-bases of the <u>s</u>-triazole series have been known since some time. 1,2 Based on the ability of the anhydro-bases 2 /which are easily accessible from the <u>s</u>-triazolium salts  $\frac{1}{2}$  to undergo ready <u>C</u>-alkylation, a novel method of synthesis of aldehydes  $\frac{1}{2}$  starting with carboxylic acids or acid chlorides and alkyl halides has been devised. The results are summarized in Table 1.

The anhydro-bases 2 were obtained by allowing to react the triazolium lodides  $\frac{1}{2}^3$  with NaH in DMF at O°C and then at r.t.  $2/R^1 = R^2 = Ph/$  is an isolable stable compound /71 %, based on diphenylacetic acid; m.p. 204 °C/. The other anhydro-bases were not isolated. After the excess NaH had been

filtered off, alkyl halides were added to their DMF solutions; the alkylations took place under evolution of heat and, after additions of aqueous KI, the compounds  $\underline{3}$  separated. The latter were in general reduced without purification with aqueous NaBH<sub>4</sub>, and the resulting compounds  $\underline{4}$  were cleaved with acid to yield the aldehydes  $\underline{5}$  as described earlier. The anhydro-base, obtained from  $\underline{1}$  /R<sup>1</sup> = Cl-C<sub>2</sub>H<sub>4</sub>, R<sup>2</sup> = H; m.p. > 320 °C, d; yield: 65 %/ suffers intramolecular C-alkylation in the course of its preparation to furnish cyclopropanecarbaldehyde as the final product.

The products were identified by comparison with authentic samples and the new intermediates of types 2 and 4 were characterized by microanalyses, IR and NAR spectra. The present procedure is a useful simple alternative to the method of Meyers. 4

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Rl	R <sup>2</sup>	x	R <sup>3</sup>	Юg	m.p.	yıeld <sup>a</sup>	m.p.	yıeld <sup>a</sup>	yıeld <sup>a,b</sup>
Н	H	Cl	Ме	I	218-20	60	90-91	53	47
H	H	Cl	Et	I	238-40	54	65-66	_	31
H	H	Cl	MeOOCCH2	Cl	232-34	55	-	-	29 <sup><b>c</b></sup>
Me	Н	OH	Me	I	243-45	40	_	-	34
Me	Me	OH	Me	I	268-70	43	82 <del>-</del> 83	34	32
C1-C2H4	H	Cl	-	-	252 <b>-</b> 54 <sup>d</sup>	49	80-81	48 <sup>e</sup>	30

- a/ The yields are throughout based on the amount of the compounds  $\mathbb{R}^1\mathbb{R}^2\text{CH-COX}$  introduced.
- b/ Isolated free aldehyde. c/ 2,4-Dinitrophenylhydrazone
- d/5-Cyclopropyl-3-methylthio-1,4-diphenyl-g-triazolium iodide
- e/5-Gyclopropyl-3-methylthio-1,4-diphenyl- $\triangle$ 3-g-triazoline

## References

- 1/ R. Grashey and M. Baumann, Angew. Chem. Internatl. Edn. 8, 133 /1969/
- 2/ G. V. Boyd and A. J. H. Summers, J. Chem. Soc. B 1971, 1648
- 3/ G. Doleschall, Tetrahedron Letters 1974, 2649
- 4/ A. I. Meyers et al., J. Org. Chem. 38, 36 /1973/, and earlier references